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# **INVESTIGATION OF SORPTION OF ENVIRONMENTAL POLLUTANTS TO VIRGIN AND AGED MICROPLASTIC**

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- increasing plastic consumption, plastic products in our environment worldwide
- plastic fragments  $< 5$  mm = Microplastics (MP)
- unknown sorption effects of environmental pollutants on microplastics

## **Creation of realistic sorption scenarios with relevant parameters**

- ➔ choice of proper materials to simulate natural occurring MP
- ➔ extensive characterization of the materials

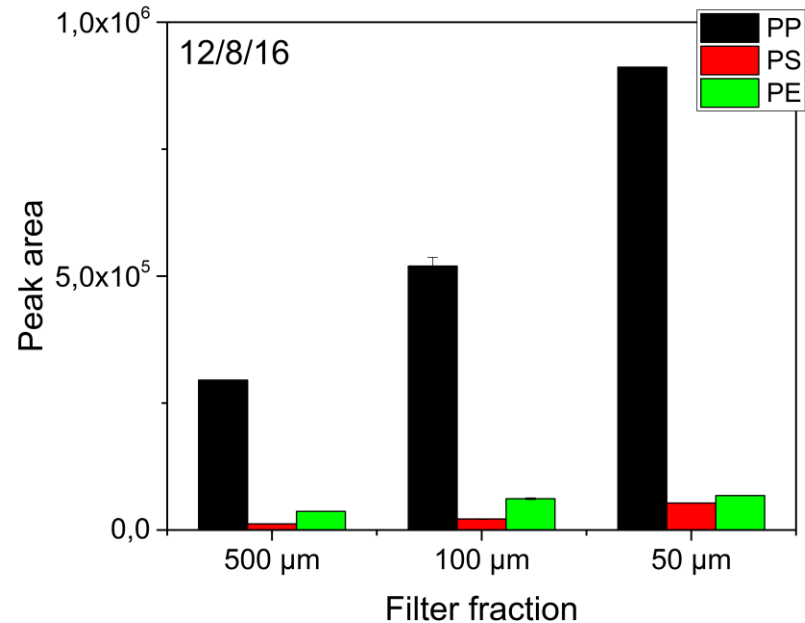
# Identification of relevant polymer types

Sampling by M. Ricking, C.G. Bannick, German Environment Agency

## detection of polymers in a sewage effluent

analytical method:  
TED-GC-MS

screening for: PS, PP, PE, PET, PA



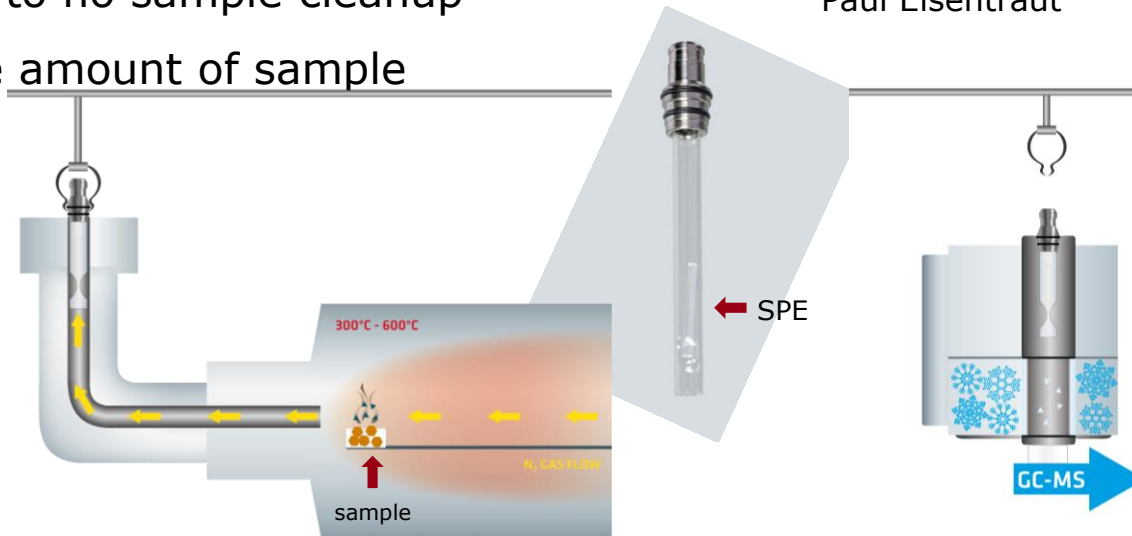
➔ **PP, PS, PE are of environmental relevance**

## Thermal Extraction Desorption-GC-MS

- identification via decomposition products
- fast analysis
- little to no sample cleanup
- large amount of sample

E. Dümichen, Fast identification of microplastics in complex environmental samples by a thermal degradation method, 2017, *Chemosphere*, 174, 572–584.

Dr. Erik Dümichen  
Paul Eisentraut



# Selection of the sorption materials

detected polymers in sewage effluent: PP, PE, PS

density below water: PE, PP  
close to water: PS  
above water: PA, PET, PVC



**PP, PS,  
PA instead of PE**

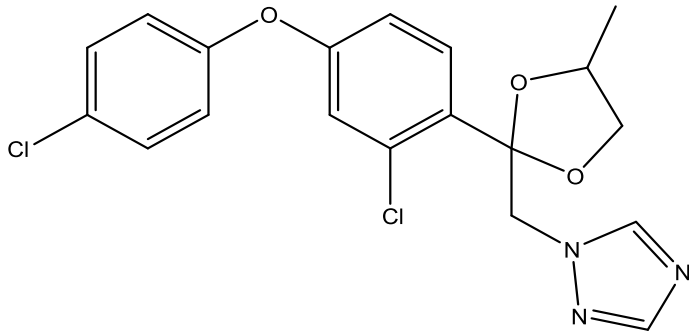
literature confirmed findings

source	sampling	analytics	Identified polymer species
Gregory 1978	Pacific Ocean (NZ)	FTIR	PE, PP
Thompson 2004	Atlantic Ocean (UK)	FTIR-microscope	PE, PP, PA, PES, Acryl
Reddy 2006	Indian Ocean (Ind)	FTIR	PU, PA, PS, PES
Frias 2010	Atlantic Ocean (P)	FTIR-microscope	PS, PE, PP
Browne 2011	various	FTIR	PES, PP, PE, PA, Acryl
Claessens 2011	Atlantic Ocean/ North Sea (B)	FTIR	PS, PP, PA, PE, PVAL
Hirai 2011	Pacific Ocean	NIR	PE, PP
Murray 2011	Atlantic Ocean (UK)	RAMAN	PE, PP
Imhof 2013	Lake Garda (I)	RAMAN	PE, PP, PS, PVC, PA
Nor 2014	Singapore	FTIR	PP, PVC, PA
Lusher 2014	North East Atlantic Ocean	RAMAN	Viscose, PES, PA

# Selection of the analyte-triacole fungicides

- one of the most used organic fungicides (market share 20 %)
- chiral
- non-polar
- perceived as persistent in soil

## Difenoconazole



## Specification

pKa	1.07
logP	4.36
water solubility	15 mg L <sup>-1</sup>
analysis	GC-MS

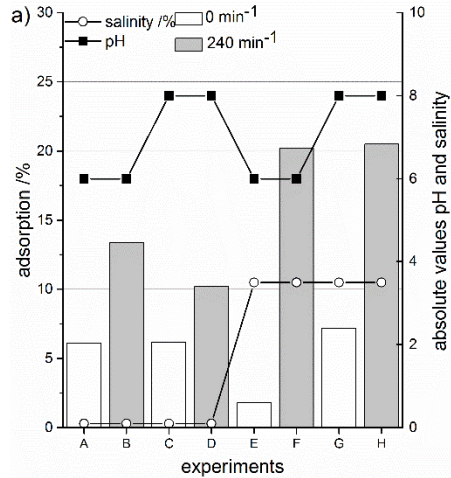
**➔ Investigation of sorption of difenoconazole on PP, PS and PA**

# Sorption experiments of difenoconazole

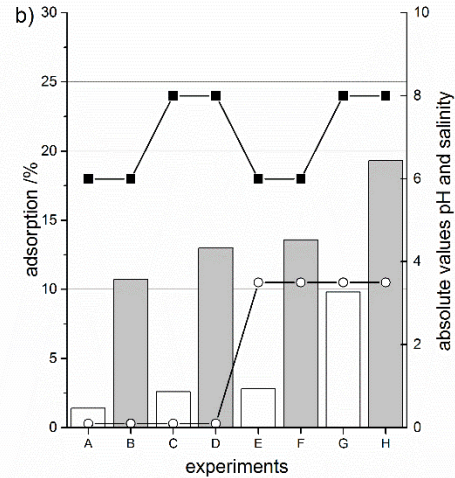
	<b>pH</b>	<b>salinity</b>	<b>agitation</b>
a	6	0.1 %	0 min <sup>-1</sup>
b	6	0.1 %	240 min <sup>-1</sup>
c	8	0.1 %	0 min <sup>-1</sup>
d	8	0.1 %	240 min <sup>-1</sup>
e	6	3.5 %	0 min <sup>-1</sup>
f	6	3.5 %	240 min <sup>-1</sup>
g	8	3.5 %	0 min <sup>-1</sup>
h	8	3.5 %	240 min <sup>-1</sup>

- full factorial design
- **1 g MP**, V=50 mL, c=0.02 mg L<sup>-1</sup>
- adsorption time **24 h**
- sorption has been **corrected with blank values** (Sorp. on glass etc.)
- 2-3 replicates per point
- quantification of difenoconazole via GC-MS

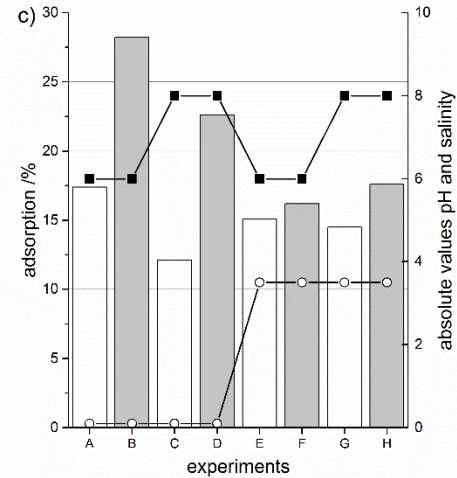
# Adsorption of Difenconazole on virgin MP



PA



PP



PS

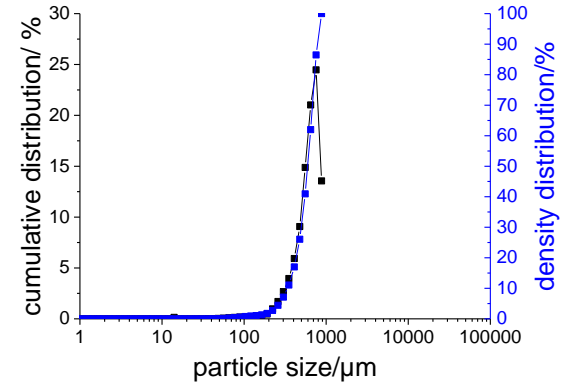
- ➔ high effect of agitation
- ➔ pH and salinity of minor importance
- ➔ highest adsorption on PS



## cryo-milling of PP



## particle size distribution

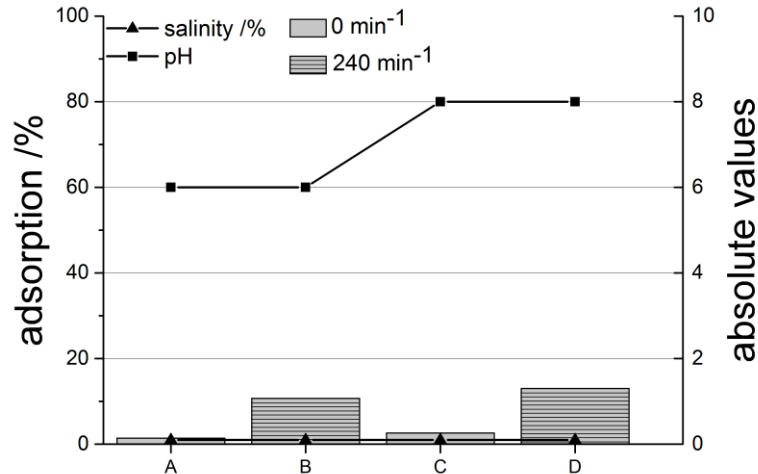


➔ enlargement of the surface:  
factor 5-14

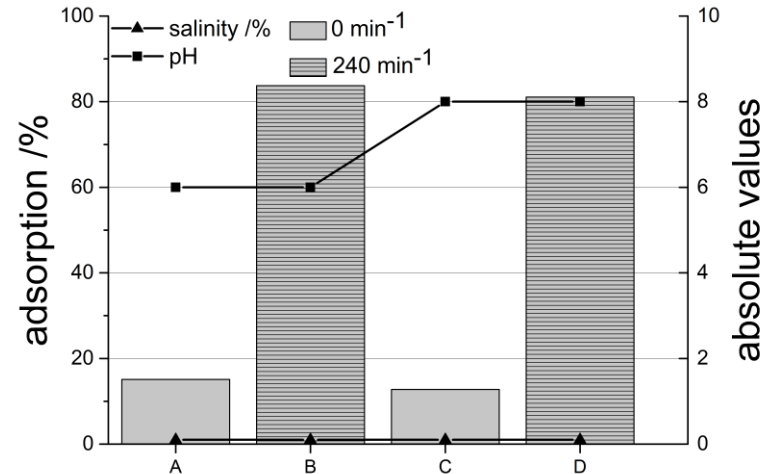
➔ environmental relevance  
of the sizes

# Influence of particle size

## virgin PP



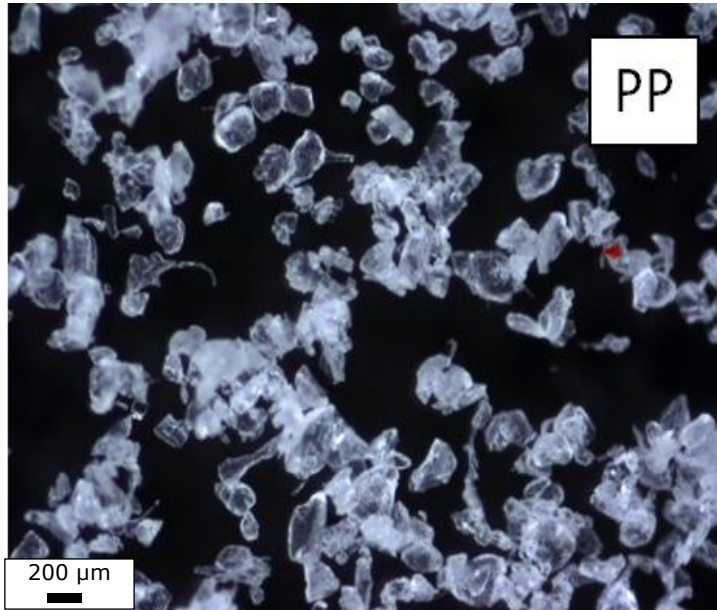
## milled PP



➔ smaller particle size leads to higher sorption effects (factor of ~4)

➔ minor effect of pH and salinity

## Photomicrograph of milled PP



- high shape inhomogeneity after milling
- spheric & non-spheric particles)
- **low  $T_{\text{glass}}$  of PP (0-20 °C)**



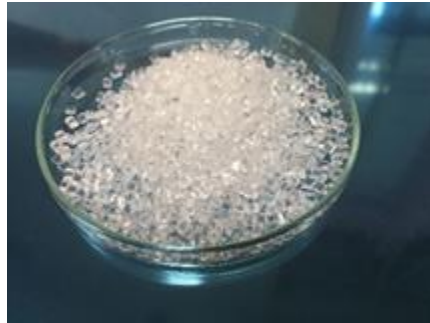
- both particle size and shape affect sorption behaviour

# Simulation of aging processes

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- PA easy to age through breaking of polymer chains
  - aging by UV-radiation time consuming
- ➔ simulation via chemical treatment

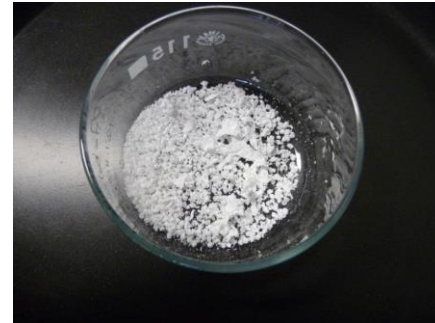
**virgin PA**



12 % HCl,  
20 % acetone

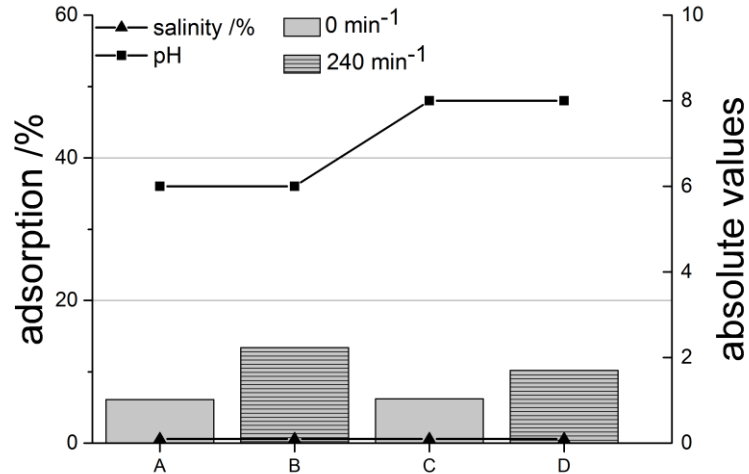


**PA flakes after  
treatment**

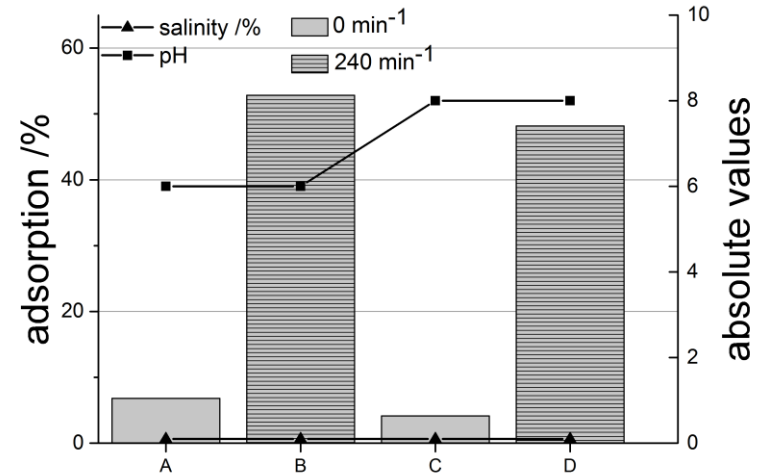


# Results of sorption on aged PA

## virgin PA



## acid treated PA

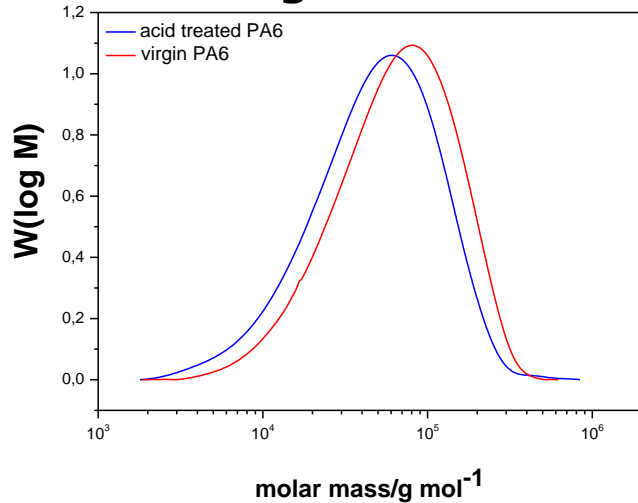


- ➔ aging leads to higher sorption effects (factor of ~3)
- ➔ small effects of pH and salinity

# Characterization of PA6

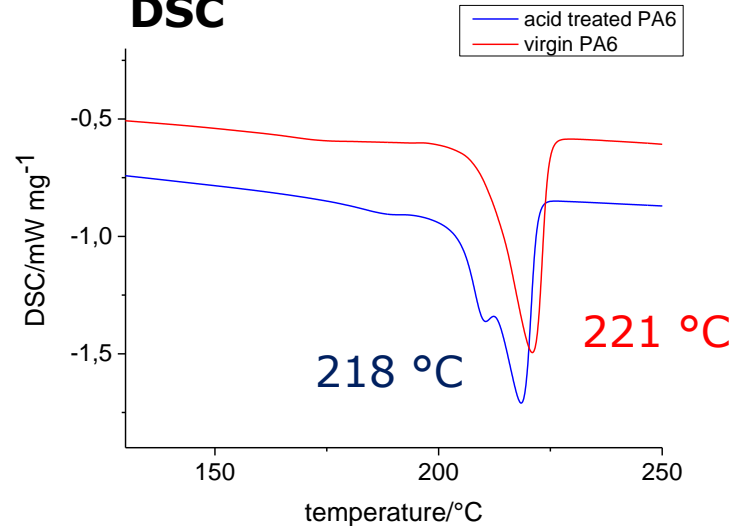
comparison of **virgin PA** & **acid treated PA6**

## GPC-elugrams



reduction of molar mass

## DSC

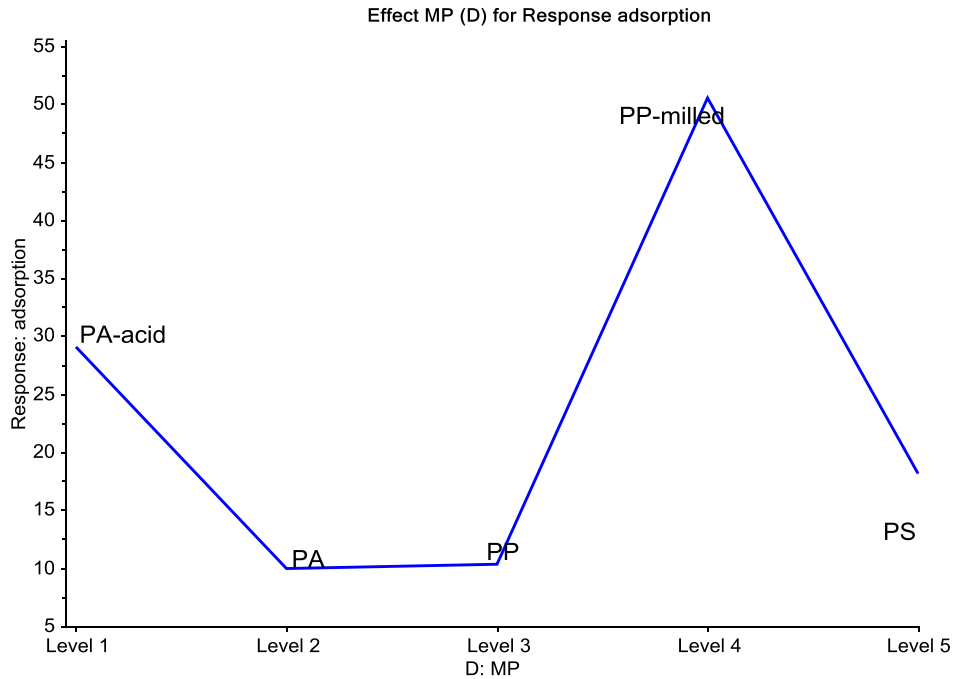


melting peak → changed crystal morphology



**more sorption, but effect less significant than particle size**

# Results of ANOVA



design „treatment“

- **milling of PP:**  
strongest effect
- **acid treatment** on  
PS and PA-acid:  
similar



results of ANOVA confirm  
results of graphic  
evaluation

- sorption of difenoconazole on all selected polymers
- effects of pH, salinity and agitation were evaluated by factorial design
- strongest effect of agitation, whereas pH and salinity are negligible
- relevant: choice of polymer type and characteristics:
  - amorphous PS sorbs stronger than semi - crystalline PP and PA
  - small PP particles sorb more difenoconazole than granulate - but difficult to characterize in model experiments
  - aged PA sorbs more than fresh, because of changed material morphology and possible increased absorption



# Acknowledgement

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Parts of this presentation are already published:

Goedecke C, Mülow-Stollin U, Hering S, Richter J, Piechotta C, et al. (2017) A First Pilot Study on the Sorption of Environmental Pollutants on Various Microplastic Materials. J Environ Anal Chem 4: 191.

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**Thank you all for your attention!**

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