

Norwegian University of Life Sciences





for Water Research



CENTRE FOR ENVIRONMENTAL RADIOACTIVITY



Norwegian Meteorological Institute



~20µm

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Radioactive Particles in the Environment -Sources and potential impact

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The speciation of radionuclides influences ecosystem transfer, biological B uptake and effects



Bulk activity concentration provides no info on processes affecting radionuclide species







Conceptual understanding of the links between: Source term - ecosystem transport - uptake - effects



Short and long term dose, impact and risk assessments

Focus: Factors contributing to uncertainties in the: Source term/deposition/speciation, Ecosystem transfer, and Effect estimates



From where?

Many sources have contributed to releases of radioactive particles Red: NMBU achieve

- Nuclear weapon tests (Kazakhstan)
- Conventional detonation of weapons (Greenland, Spain)
- > Nuclear reactor explosions and fires (Ukraine, UK, Canada)
- > Accidents with reactor driven vehicles: satellites, submarine accidents (Russia)
- Effluents from nuclear installations (UK, France, USA, Russia, Sweden)
- > Leaching from dumped nuclear material (Kara Sea, Barents Sea)
- Uranium mining and tailing (Central Asia)
- Use of DU ammunition (Kosovo, Kuwait)

Radioactive particles containing a series of radionuclides and metals













Radioactive particles released during "all" types of severe nuclear events. The source determines the composition, the release scenarios dictate particle properties



Advanced techniques available for particle characterization – state-of-the-art

- Hot spots/heterogeneities: digital autoradiography and sample splitting gamma measurements
- Particle size, surface structure and elemental composition: ESEM with XRMA, TEM with XRMA
- Subsurface/volume elemental composition: SR-based 2D μ-XRF (fluorescens)
- Oxidation state determination: SR-based 2D µ-XANES (micro X-ray absorption near edge structure spectrometry
- Crystallographic structure: SR-based µ-XRD (micro X-ray diffraction)
- 3D elemental distribution: Confocal µ-XRF and nano- XRF, TOF-SIMS
- 3D structure distribution: Tomographic µ-XRD and nano XRD Utilizing new nanobeamlines (ESRF, PETRA)
- Source identification: Isotope or atom ratios by MS techniques (ICP-MS, AMS)
- Weathering and mobilisation potential: Leaching experiments combined with particle characterization techniques



Info: Particle size distributions, surface elemental distributions, identifying single U particles: Autoradiography – SEM/BEI mode



INFO: 2D and 3D elemental distribution by Synchrotron radiationµ-XRF mapping. Chernobyl particles containing a series of radionuclides





Chernobyl particles: inclusion of Ru+Mo Corresponding distributions: U, Zr, Sr



INFO: oxidation states of U in a Chernobyl U particle by synchrotron radiation µ**-XANES (ESRF)**



INFO: 2D and 3 D elemental distributions by synchrotron radiation based XRF mapping of a particle from underground nuclear weapon tests, Degelen Mountain, Semipalatinsk



Inside:







3D Density distribution – absorption tomography -Particle from under ground test, Semipalatinsk, Kazahkstan

Red-dense Light yellowless dense/ hollows

Single absorption tomogram through spherical particle





Concept: Particle weathering and remobilisation



Chernobyl case in short: particle weathering and release of Cs-137 and Sr-90: Cs-137 fixed to soils, Sr-90 mobilised and released to the environment



Abiotic: mimicking human stomach juice Linking particle characteristics to potential bioavailability (OC Lind))



U

- Low and moderate temperature events → relatively high solubility
- Increasing ox state of U increasing solubility (fire)
- Low solubility for particles from:
 - High temperature events under reducing conditions (e.g., formation if U-O-Zr, U-O-C)
 - Formation of refractory oxides



Do they matter? CASE: The Chernobyl accident: 3-4 tons of fuel released Explosion – inert, Fire - oxidised



Does it matter? Uptake and accumulation of µm-nm particles in organisms EU COMET - RATE



Dounrey particles given to Blue

Mussels as food, retained in the

gut and deposited in tissues

Radioactive particles retained in grazing goats, Incorporated in GI tissues

Severe skin dose about mGy/hr, unevenly distrib Effects: Increase Comet DNA % and increase in micronucleus frequency the haemolymph; nontargeted effects



⁶⁰Co NPs uptake in reproductive organs of earthworm Coutris et al, 2012





ESRF News July 2015 Beauty of science: Co nanoparticle uptake in nematodes (CERAD D. Brede)





Co NP (5-10 nm) given to synchronized C. elegans (1 mm). Internal distribution: CT, 2D XRF tomography at the ESRF ID16A synchrotron beamline. Endpoint: Reproduction failure at the highest concentrations

First time 2 and 3D distribution of NPs are imaged in an intact (50x1000 μm) organism at 30 nanometer resolution

Looking for possible transmembrane transport of NP



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Retention of radioactive particles in biota in the field, Palomares RATE results (UoSeville, CIEMAT, NMBU)



Objectives: Demonstrate radioactive particle retention in biota collected in the field and characterize the exposure
Methods: gamma spectrometry, nano-CT, synchrotron 2D/3D μ-XRF/μ-XRD



Fragment of shell of Palomares snail with embedded U/Pu particle



Snails are exposed to radioactive U/Pu particles originating from the Palomares nuclear accident as well as NORM particles through ingestion of soils and particles can be retained and embedded in shells of soil dwelling snails.

In the field, Pu particles are taken up by organisms such as moskoxen from Thule, Greenland (P.Roos), and hare and snails from Palomares, Spain - calls for better exposure characterization in impact assessments of particle contaminated areas



Do they matter?

Impact of deposited radionuclide species (mobile species and inert particles) on ecosystem transfer



Source term		Transport	Biological → uptake	Dose-assessment
Impact of	Speciation	K _d	CF	mSV
Mobile species	High load of mobile species	Low Increase <i>f(t)</i> when transformed into reactive species	High in fish Low in benthic Decrease <i>f(t)</i> in fish, Increase in benthic for reactive species	Underestimated short- term dose-assessment Overestimated long- term dose-assessment for reactive species
Particles	High load of inert species	Very high Decrease <i>f(t)</i> due to weathering	Low in fish High in benthic Increase <i>f(t)</i> in fish Decrease <i>f(t)</i> in benthic	Overestimated short- term I dose-assessment Underestimated long- term dose-assessment for mobile species

Transfer coefficients (Kd, CF, BCF, TF, TC) are based on assumption on equilibrium – should be describes as time functions



Take home message



- If refractory radionuclides are released expect particles Challenge: to find and extract single particles
- Many sources particles with different properties Challenge: to link specific particles to specific source and release scenarios
- Particle characteristics advances techniques needed

Challenge: access to advanced platforms and relevant competence

• Retention in soils/sediments – apparent Kd is high

Challenge: implement Kd as a time function (e.g., particle kinetics)

- Particle weathering rates remobilization of associated nuclides Challenge: link kinetics to particle properties
- Uptake/accumulation/effect in organisms uneven dose distribution Challenge: further development of micro/nano dosimetry and point source effects needed



So particles do matter!

Thank you

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