

Use of mediated electrochemical analysis and high field FTICR mass spectrometry to explain humic acid fractionation upon sorption to redox inert sorbents

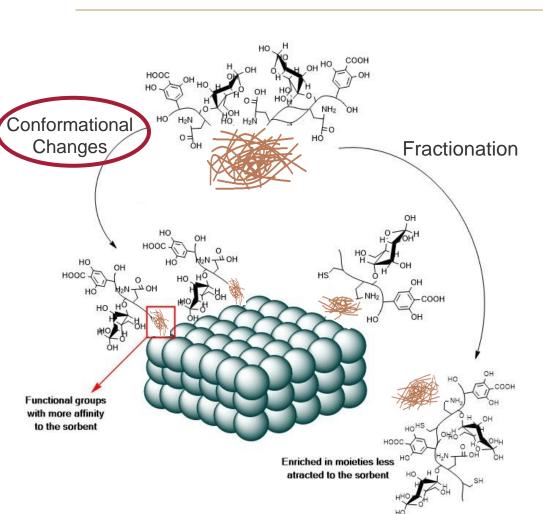
E. Subdiaga^a; S. Orsetti^a; M. Harir^{b,c}; P. Schmitt-Kopplin^{b,c}; S. B. Haderlein^a

^aEberhard Karls Universität Tübingen, Center for Applied Geosciences, Environmental Mineralogy and Chemistry Group, Hölderlinstr. 12, 72076, Tübingen, Germany.

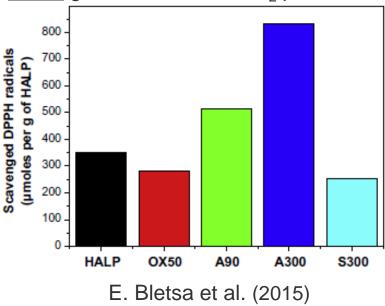
^bResearch Unit Analytical Biogeochemistry (BGC), Helmholtz Zentrum München, German Research Center for Environmental Health, Ingolstädter Landstrasse 1, 85764 Neuherberg, Germany.

^cChair of Analytical Food Chemistry, Technische Universität München, 85354 Freising, Germany.

16th ICCE Conference – Oslo, Norway – 06/20/2017



HALP: Humic Acid Like Polycondensate A300: grafted HALP on SiO₂ particles



A300 → 300% higher antioxidant activity than only HALP

Conformational arrangements might trigger changes in HA properties.

^{2 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen

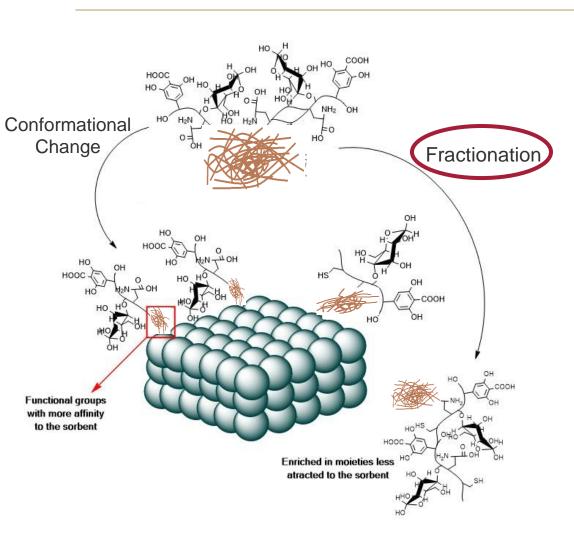


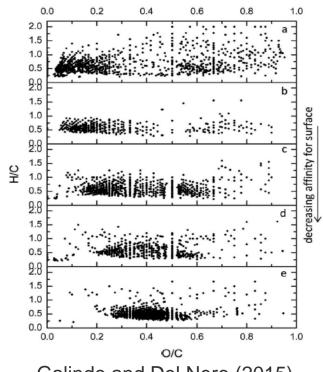
Fa

Faculty of Science

Center for Applied Geosciences

Overview





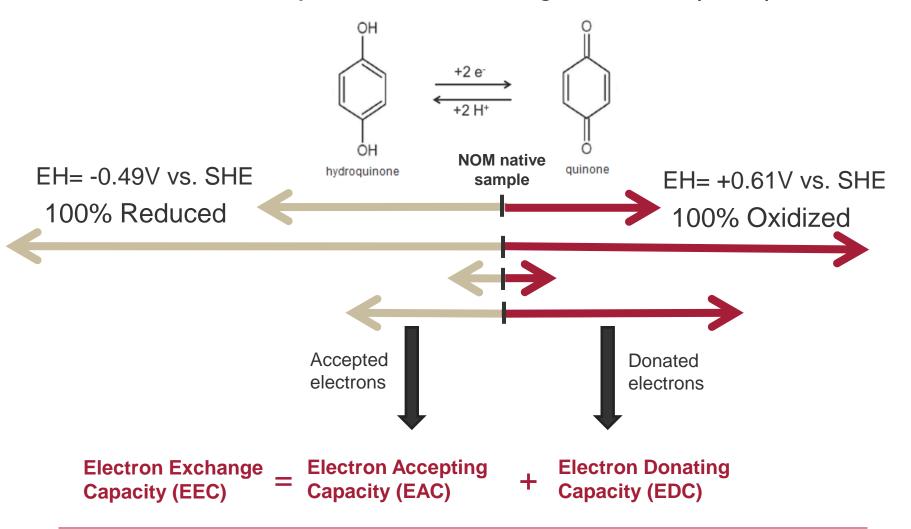
Galindo and Del Nero (2015)

 Enrichment of highly oxygen functionalized aromatic and aliphatic molecules on Al₂O₃ surface

Fractionation is caused by Humic Acid (HA) compositional heterogeneity

^{3 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen

Electrochemical Properties of Natural Organic Matter (NOM)



^{4 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen

 Does sorption of HA onto redox inert surfaces trigger changes in HA chemical properties (in the absence of electron transfer)?... and to which extent?

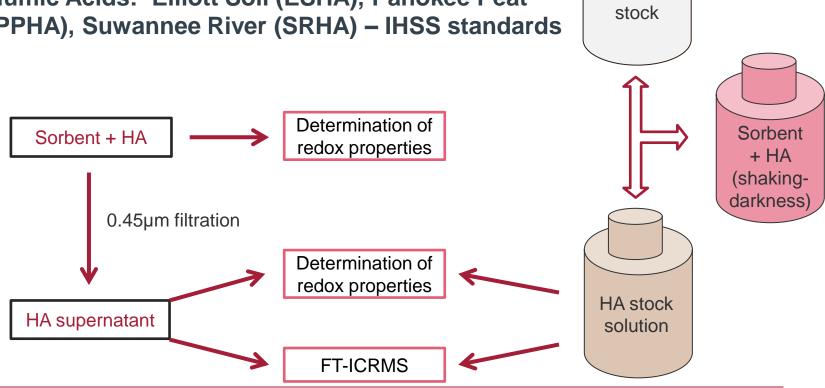


 Which sorption phenomena (fractionation, conformational arrangements, etc.) lead to changes in HA properties?

sorbent



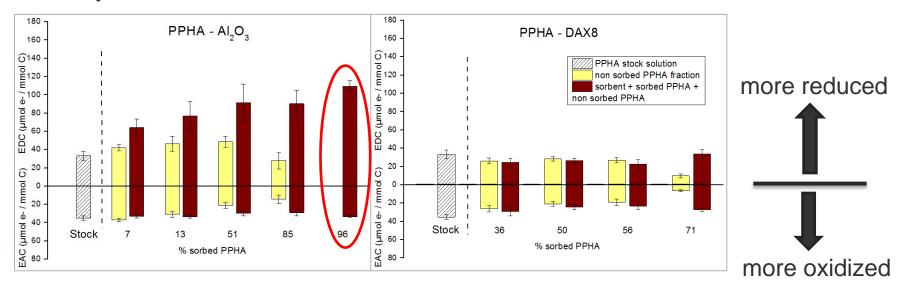
- Anoxic conditions (O₂ < 0.1ppm)
- pH 7 (no buffer use 0.1M KCl medium)
- Sorbents: Aluminium Oxide (Al₂O₃), resin DAX-8
- Humic Acids: Elliott Soil (ESHA), Pahokee Peat (PPHA), Suwannee River (SRHA) – IHSS standards



^{6 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen



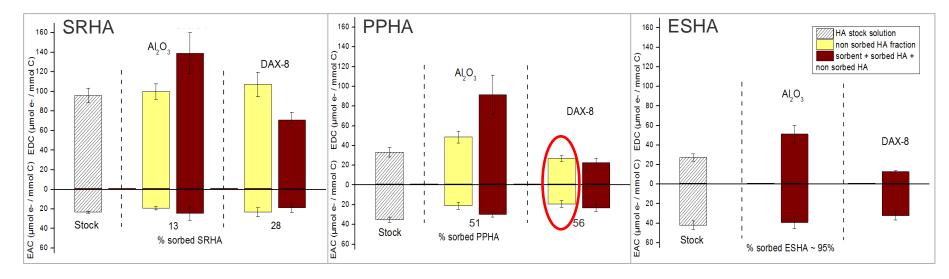
Sorption level effect?



- Al₂O₃:
 <u>↑</u> % sorbed HA → up to 300% higher EDC whole suspension
- DAX-8:
 ↑ sorbed HA → ~ 50% lower EDC whole suspension
- Optical analysis (UV EEM fluorescence) did not provide clear indication of HA fractionation



Sorption mechanism? \rightarrow Al₂O₃ vs. DAX-8?



- Al₂O₃: EDC suspension > EDC HA stock solution
- DAX-8: EDC suspension < EDC HA stock solution
- Above ~ 50% sorbed HA, EAC & EDC decreased in supernatants

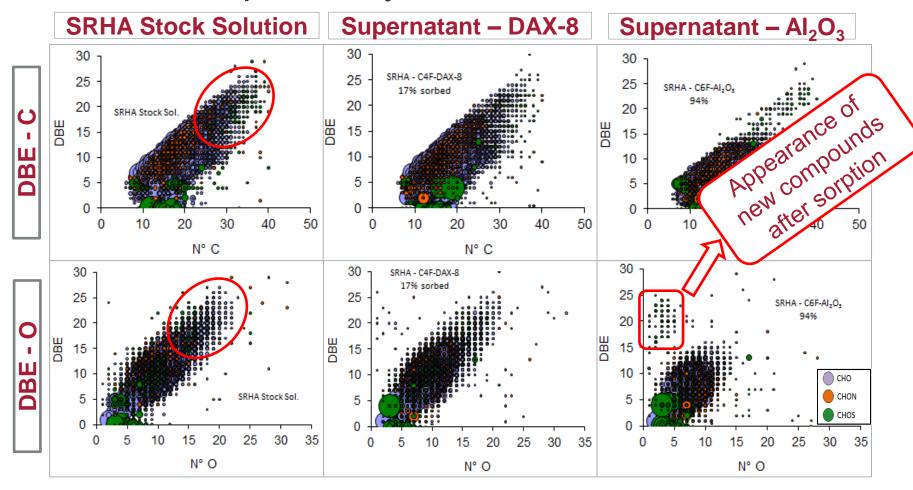




Faculty of Science

Center for Applied Geosciences Double Bound Equivalent (DBE)

FT-ICR mass spectrometry results

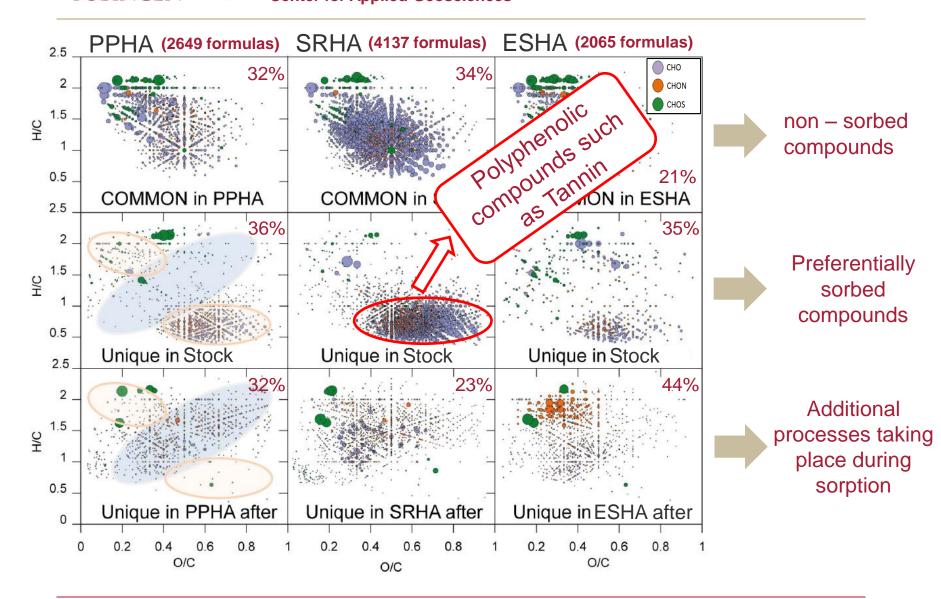


 Al_2O_3 : significant difference at N°C> ~25 and DBE > ~15

^{9 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen

Faculty of Science Center for Applied Geosciences

Existence Analysis



^{10 |} Use of mediated electrochemical analysis and FT-ICRMS to explain humic acid fractionation upon sorption to redox inert sorbents © 2017 - University of Tuebingen



- Fractionation of HA upon sorption was confirmed by mediated electrochemical and FT-ICRMS analysis.
- No significant selective sorption of HA at DAX-8.
- Strong selective fractionation of HA components occur upon sorption at Al₂O₃. Poly-phenolic (Tannin) like seem to be leading preferentially sorbed compounds.
- Investigate further processes occurring upon HA sorption at polar minerals.
- To study systems where the sorbent is redox active (clays and iron minerals).

- Dr. Michael Sander (ETH-Zurich) for great support and fruitful discussion on electrochemical set-up.
- Monika Hertel, Bernice Nisch, Ellen Struve (Uni. Tübingen).
- Sharmishtha Jindal and Michael Trumpp for experimental support.

Thanks for your attention!