



Monitoring of phosphorous fractions

Understanding the hydro-geochemical processes governing mobilization and transfer of phosphorous in an agricultural watershed in north-eastern China



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Water quality in China

- Rapid economic growth in the last 30 years
 - Rising living standards - High consumption
 - Urbanization
 - Population increase
 - Increased agriculture and industries
 - Increased water extraction and **pollution**
- Eutrophication - Most critical problem facing lakes and reservoirs in China
- Over **58%** of the lakes are eutrophic/hypertrophic (*Chai et. al, 2006*)



SinoTropia project

- SinoTropia is a Sino-Norwegian trans-disciplinary project focusing on understanding eutrophication in China.

-Assessing the impact of changes in environmental pressures on **mobilization, transport, fate and impact** of phosphate fractions to the Yuqiao reservoir in Tianjin, China.

- Why Sinotropia?

- Limited knowledge on mobilization, transport and fate of phosphate

-The need for site specific abatement actions

Aim of the study

- Access sources, mobilization and transport of phosphates to the Yuqiao reservoir
 - P fractionation
 - DGTs

Drivers of Eutrophication in China

Animal Husbandry - Manure



Sewage/effluent



Agriculture - Fertilizers



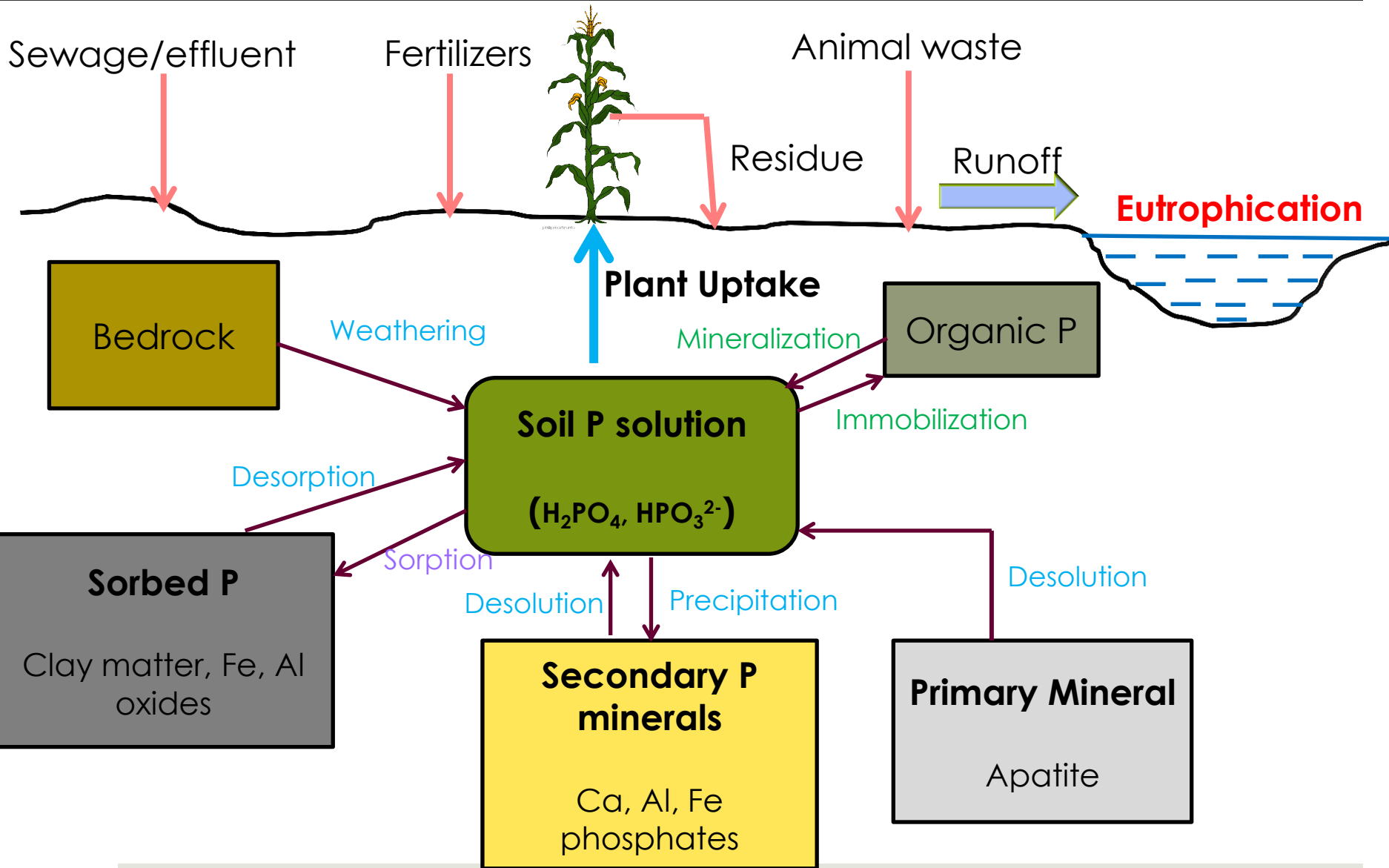
Soil - Impeable clay



Monsoon climate



Phosphorous



Phosphorous fractionation

Total fraction	Total P (TP) by digestion			
Filtration (0.45µm)	Particulate P (PP) (on filter)		Dissolved P (TDP) (filtrate)	
Fractions	Particulate Organic P	Particulate Inorganic P	Dissolved Organic P	Dissolved Inorganic P
Denotation	POP	PIP	DOP	DIP
Bioavailability	Low	Low	Medium	High

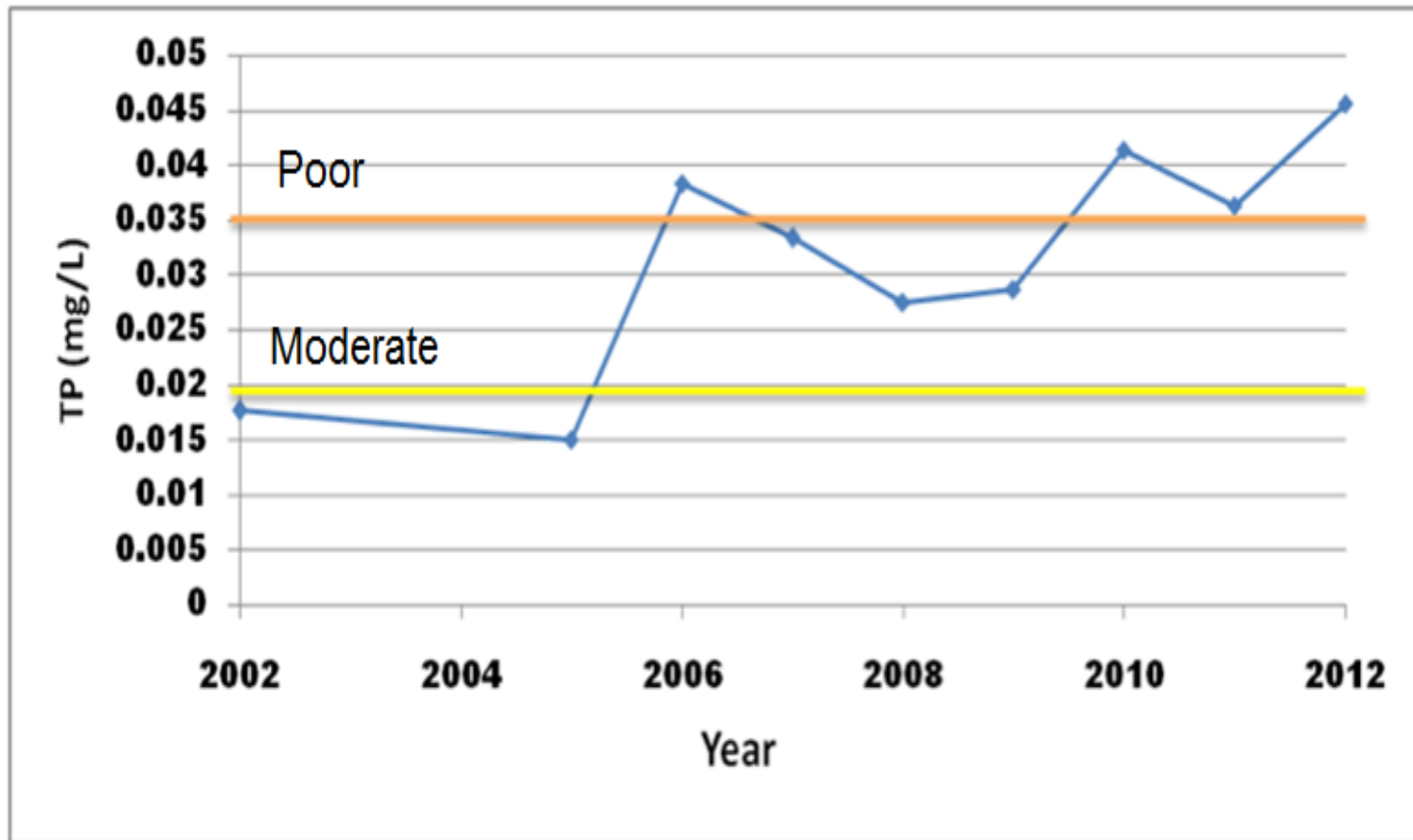
Yuqiao reservoir



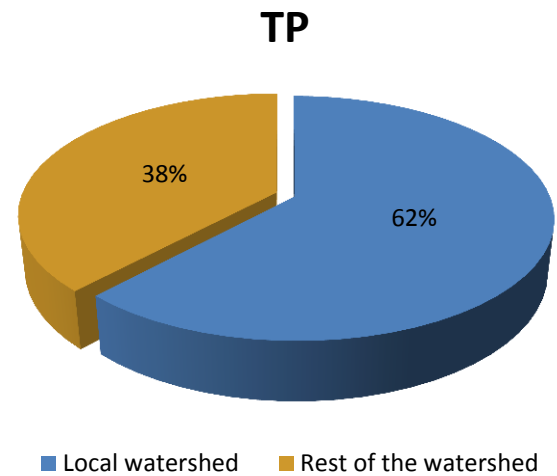
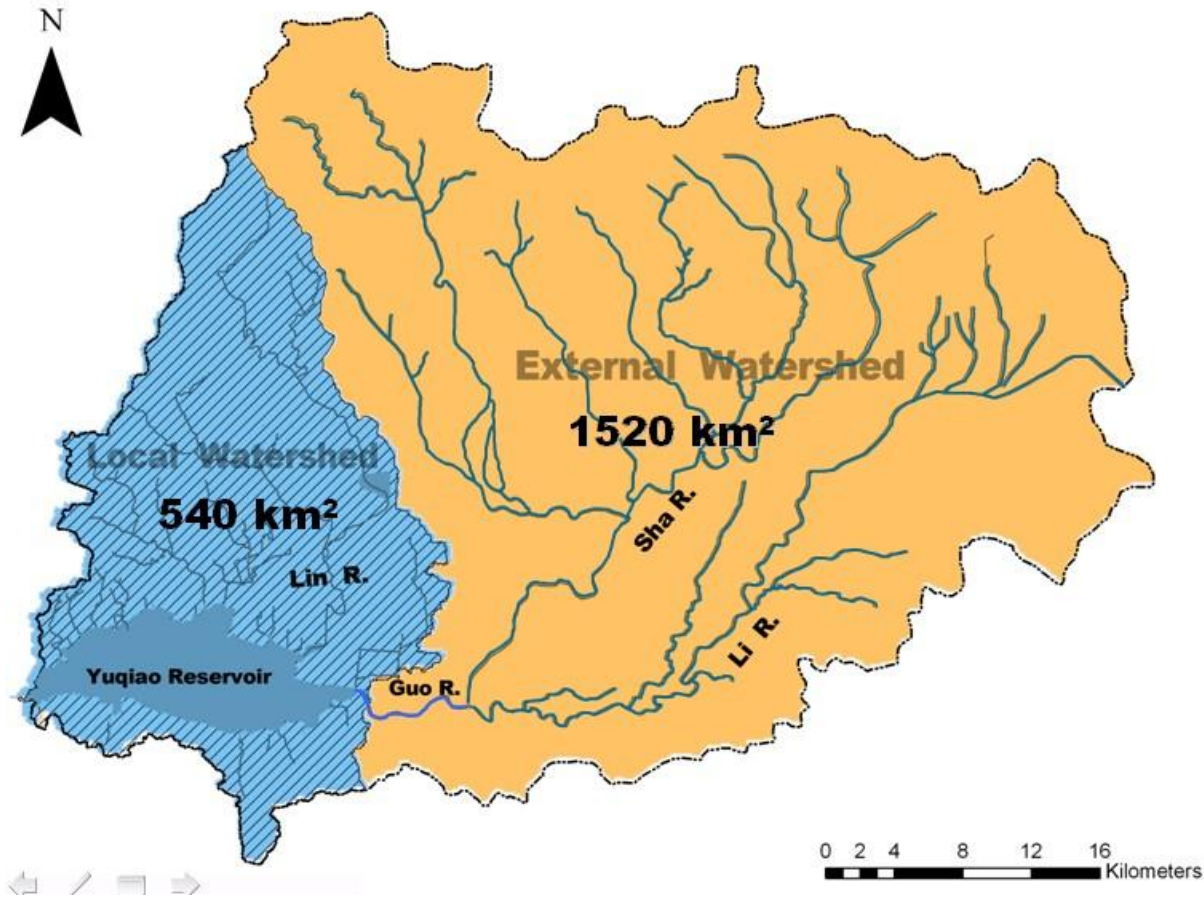
- Source of drinking water for over 6 million people
- Facing eutrophication challenges

Nutrient level – Yuqiao reservoir

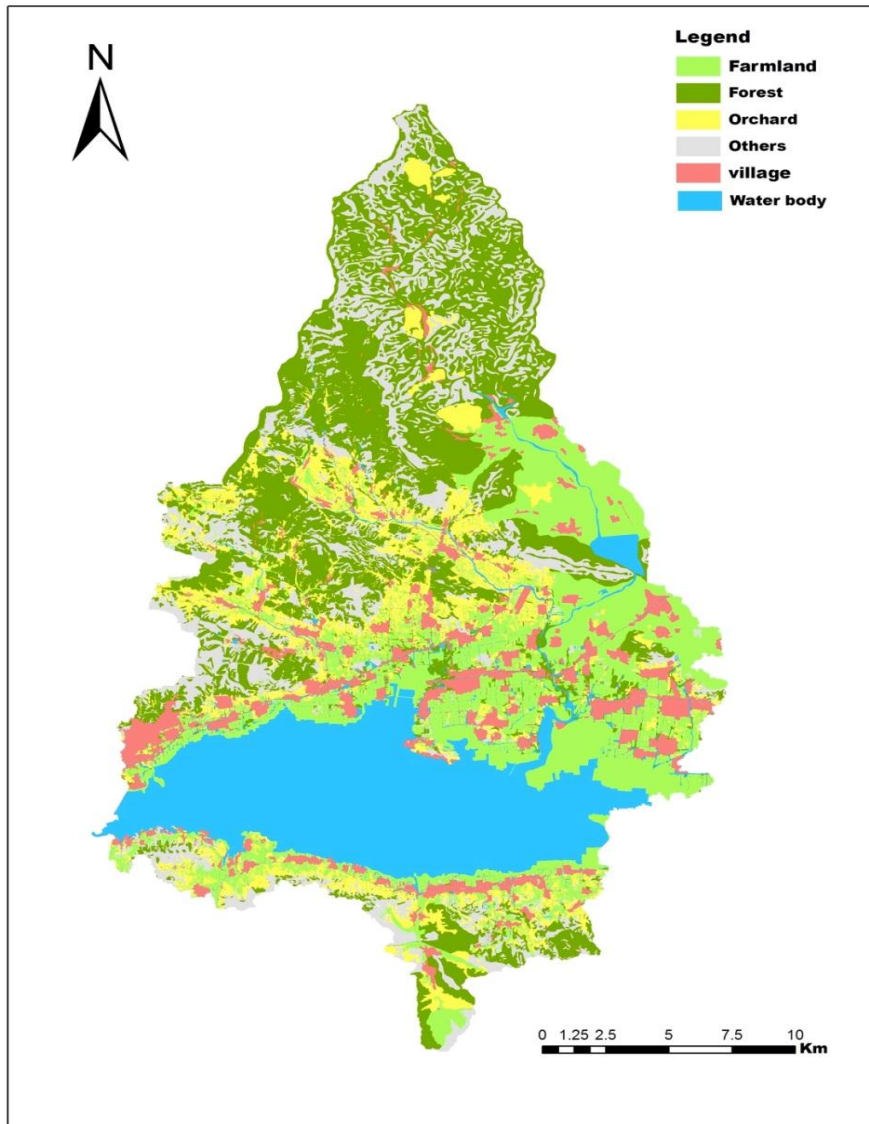
- Reservoir – Experience algae bloom (Summer/Fall)
- Increasing trend of eutrophication



Site description



Land Use



Animal husbandry and aquaculture



Sampling – Two types

Water sampling

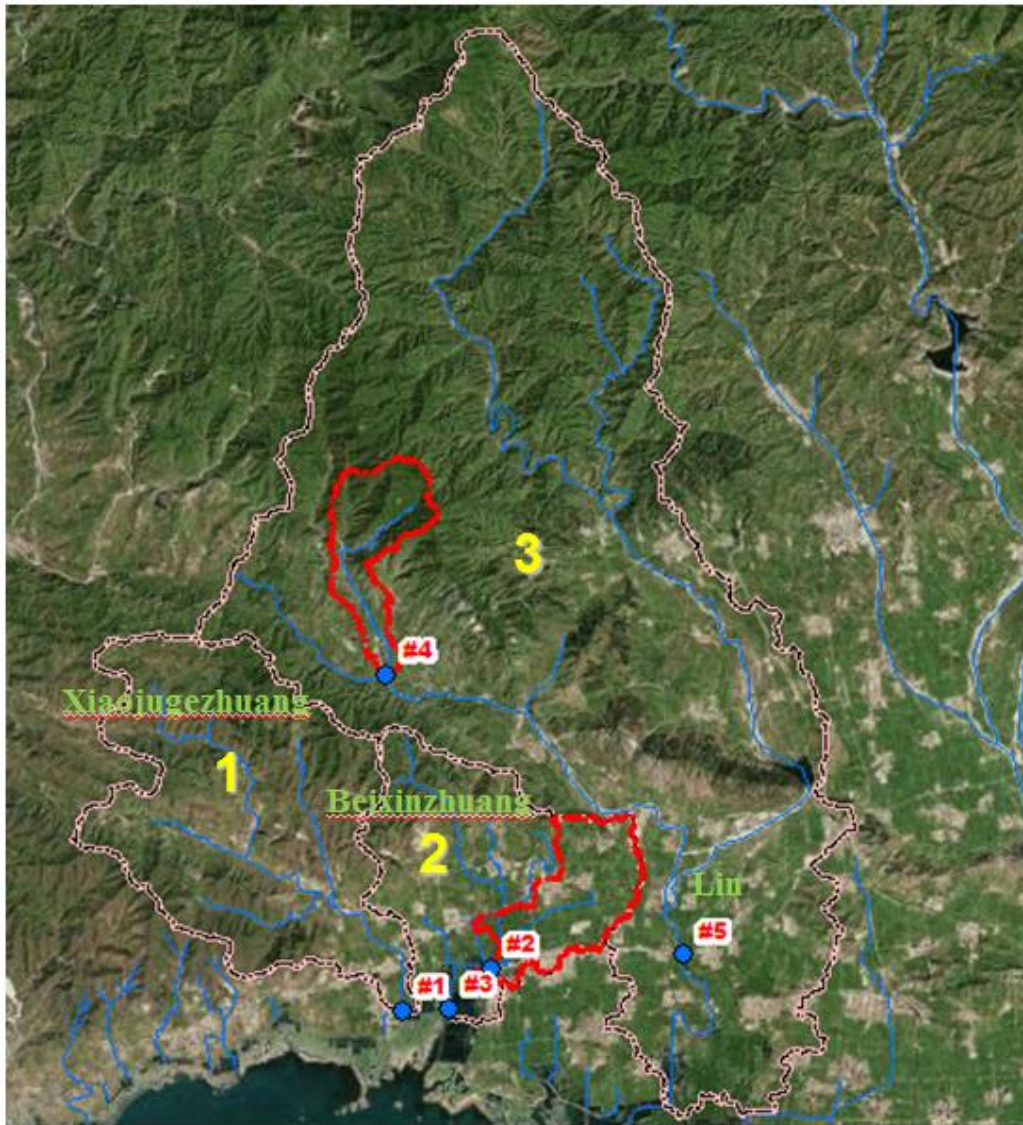
- Synoptic and Episodes studies (2012 - 2013)
- Three (3) rivers – 5 points
- 348 samples collected

DGT sampling

- Done during wet months (July – September 2014)
- Three (3) rivers – 5 points
- 57 samples collected



Sampling sites - Rivers



- Three (3) river basins
- Five (5) sampling points
- Catchments by **land use**
-proxy for P fraction distribution
- Sampled catchments
 - Forest (**#1**)
 - Farmland (**#2**)
 - Orchard (**#3**)
 - Mixed 2- Mountain (**#4**)
 - Mixed 1- Watershed (**#5**)

Sampling sites - Reservoir and fish ponds



Yuqiao Reservoir



Fish pond

Analysis methods

▣ Water analysis:

- Parameters measured: - pH, Alkalinity, Cations, Anions, TP, PO₄, TSS

▣ Particle characterization:

- Organic and Inorganic phosphates (Loss of Ignition)
- Minerology (XRD)
- Elemental composition: Microwave digestion (68% HNO₃ only)
ICP-OES (Na, Mg, Al, K, Ca, Mn, Fe, Si and P)

▣ DGT analysis

- DIP and TDP (Molybdate Blue Method and ICP-MS)

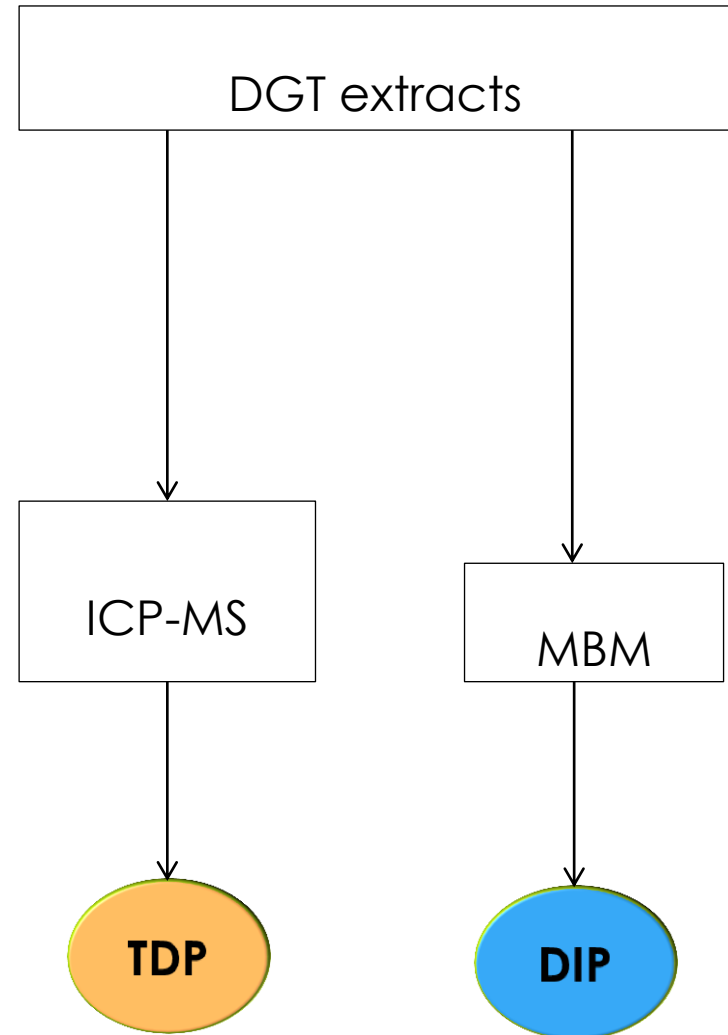
DGT extraction and P analysis

- Disassemble DGT and remove resin gel
- Place resin gel in tube and add H_2SO_4
 - Ferrihydrite dissolves and phosphate is released

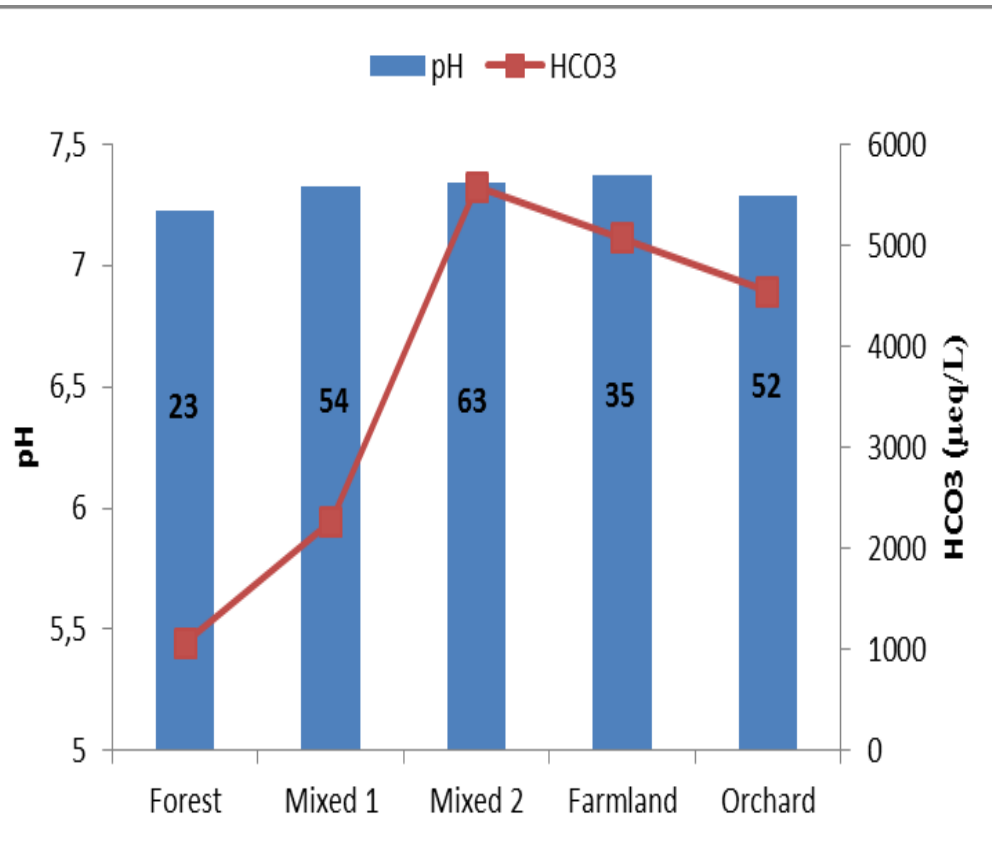
Two P fractions:

- Total Dissolved P (TDP)
- Dissolved Inorganic P (DIP)

Note: Dissolved Organic P (TDP-DIP)

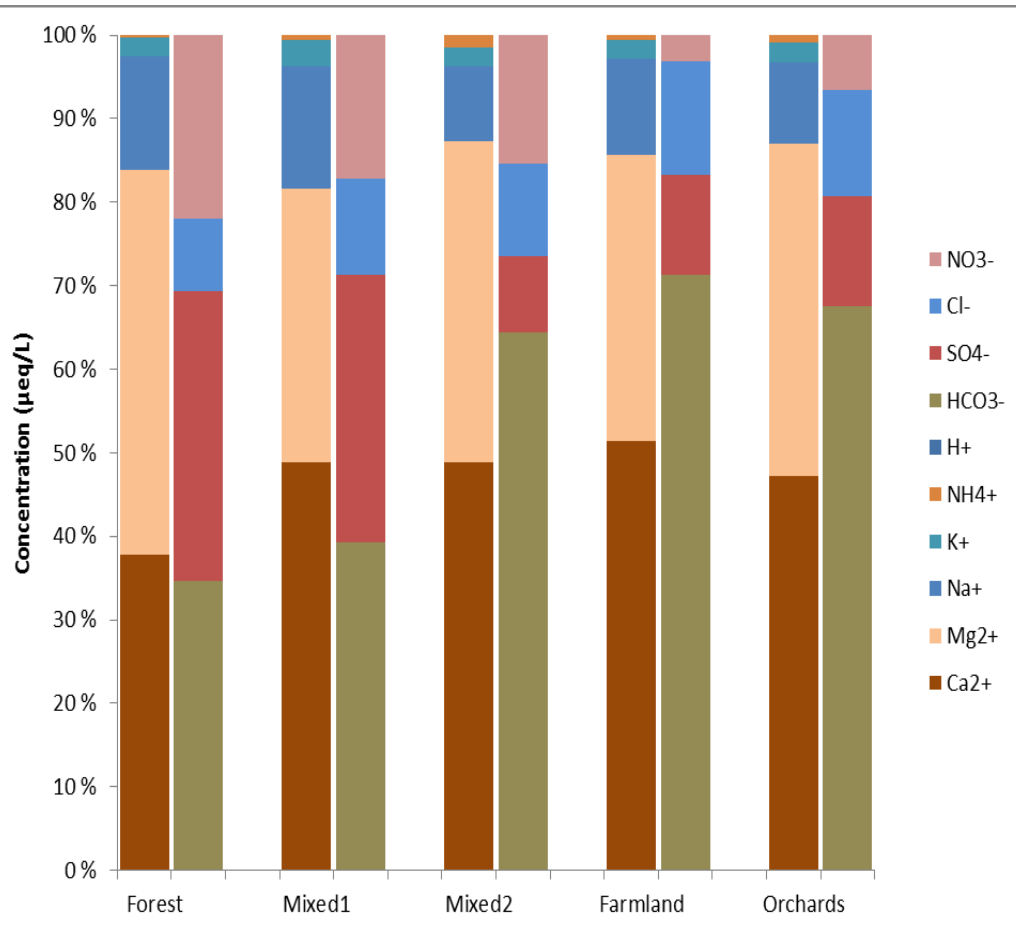


Water - pH and Alkalinity



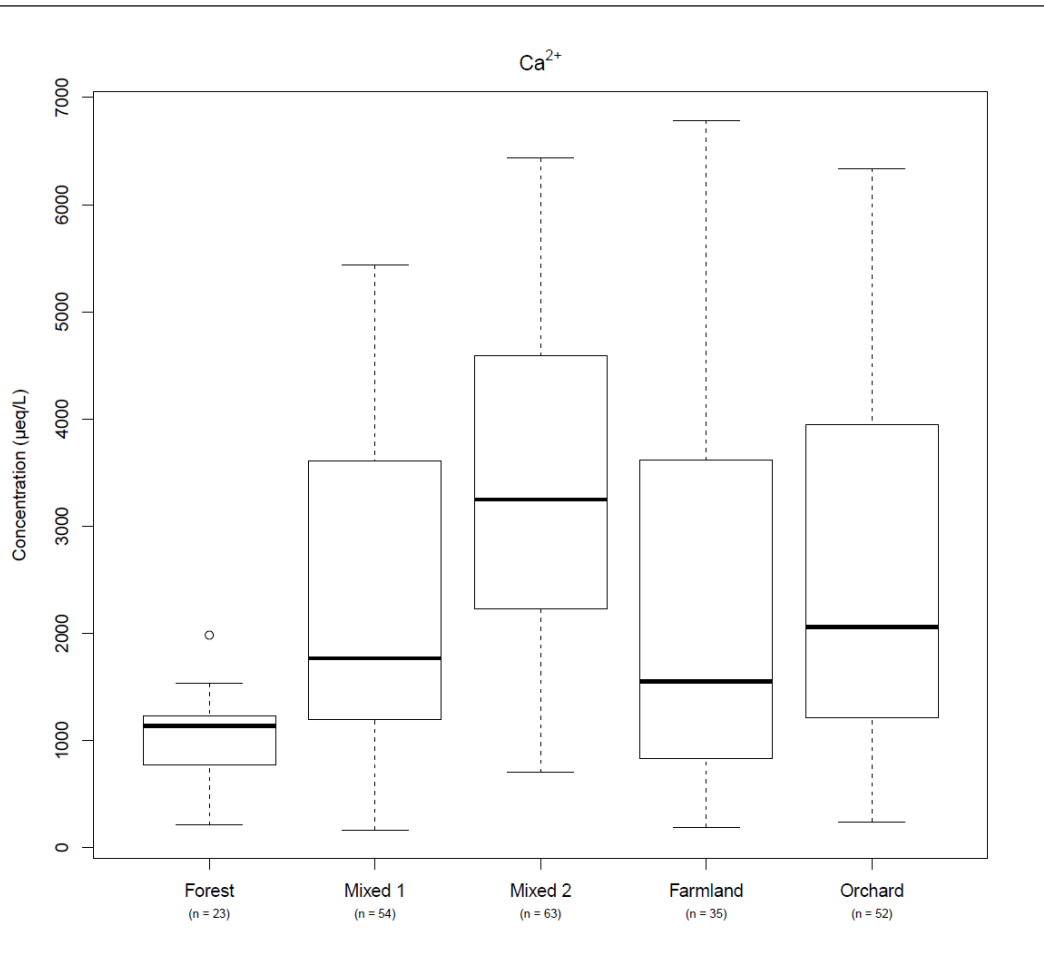
- pH range **7 – 7.5**
- P governed by Ca precipitation
- Buffering by carbonates rocks, liming and manure
- Difference in amount of bicarbonates
- Large difference in buffering

Water - Major cations and anions



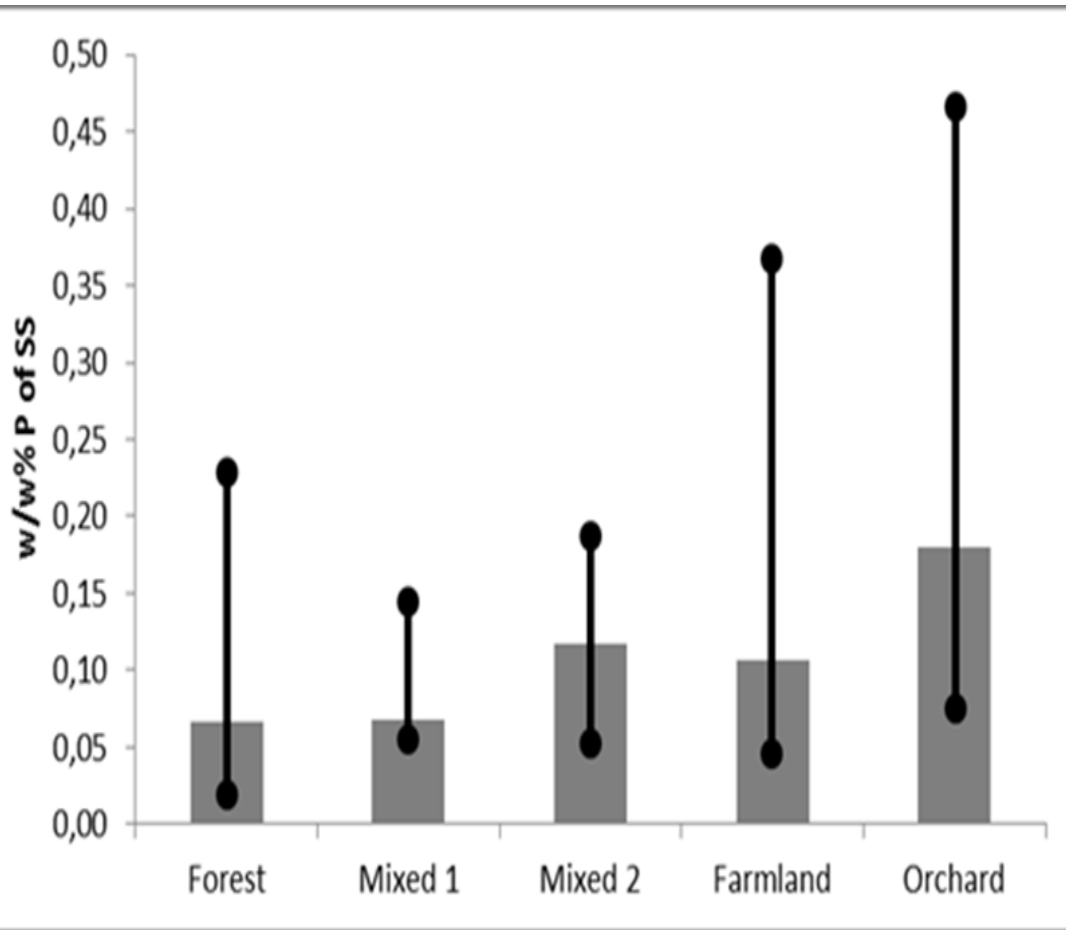
- Difference in concentration strength
- Large charge balance discrepancy in farmland
- Major cations: **Ca²⁺** and **Mg²⁺**
- Major anion: **HCO₃⁻** except in the forest with **SO₄²⁻**
- Surprise low **K⁺** and **NO₃⁻** in farmlands and Orchards

Water -Cations across land use



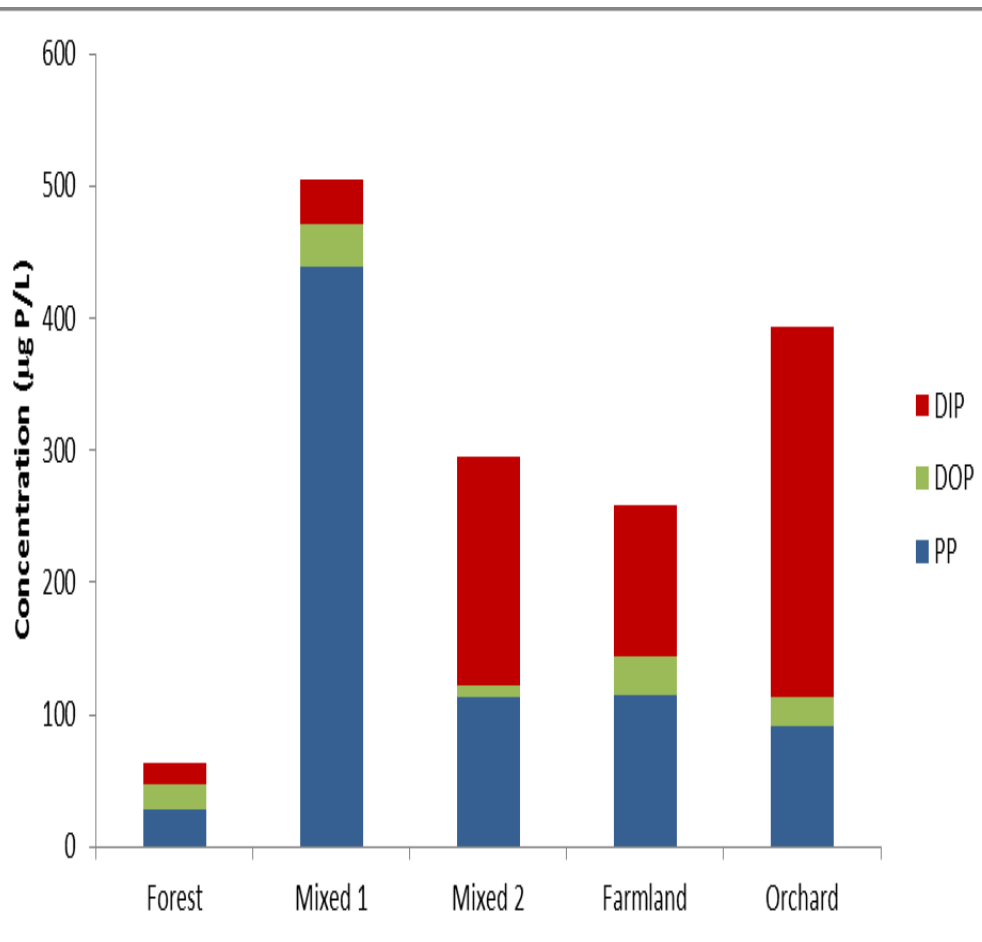
- **K⁺**: Basically same for landuses
 - Addition as fertilizer does not results in high K⁺ concentration in runoffs
 - No surplus in nutrient needs
- **Ca²⁺ and Mg²⁺**: Conc. variations
 - Liming
 - Weathering
 - Ion exchange with soil and
 - Management practises

Suspended matter



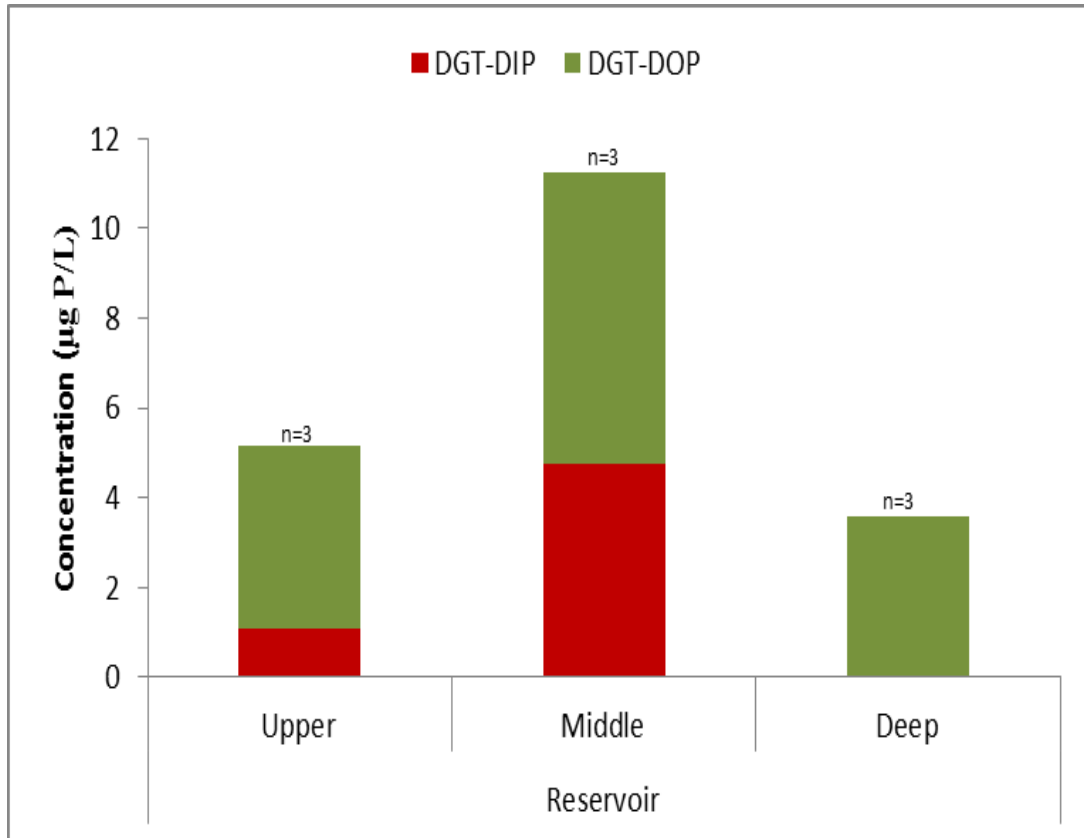
- Loading variations
 - Land Use
 - River velocity
 - Soil
 - Topography
- High loading in **Farmland** and **Orchards**
 - Tilling
 - Soil erosion
 - Management practises

Phosphorous fractions



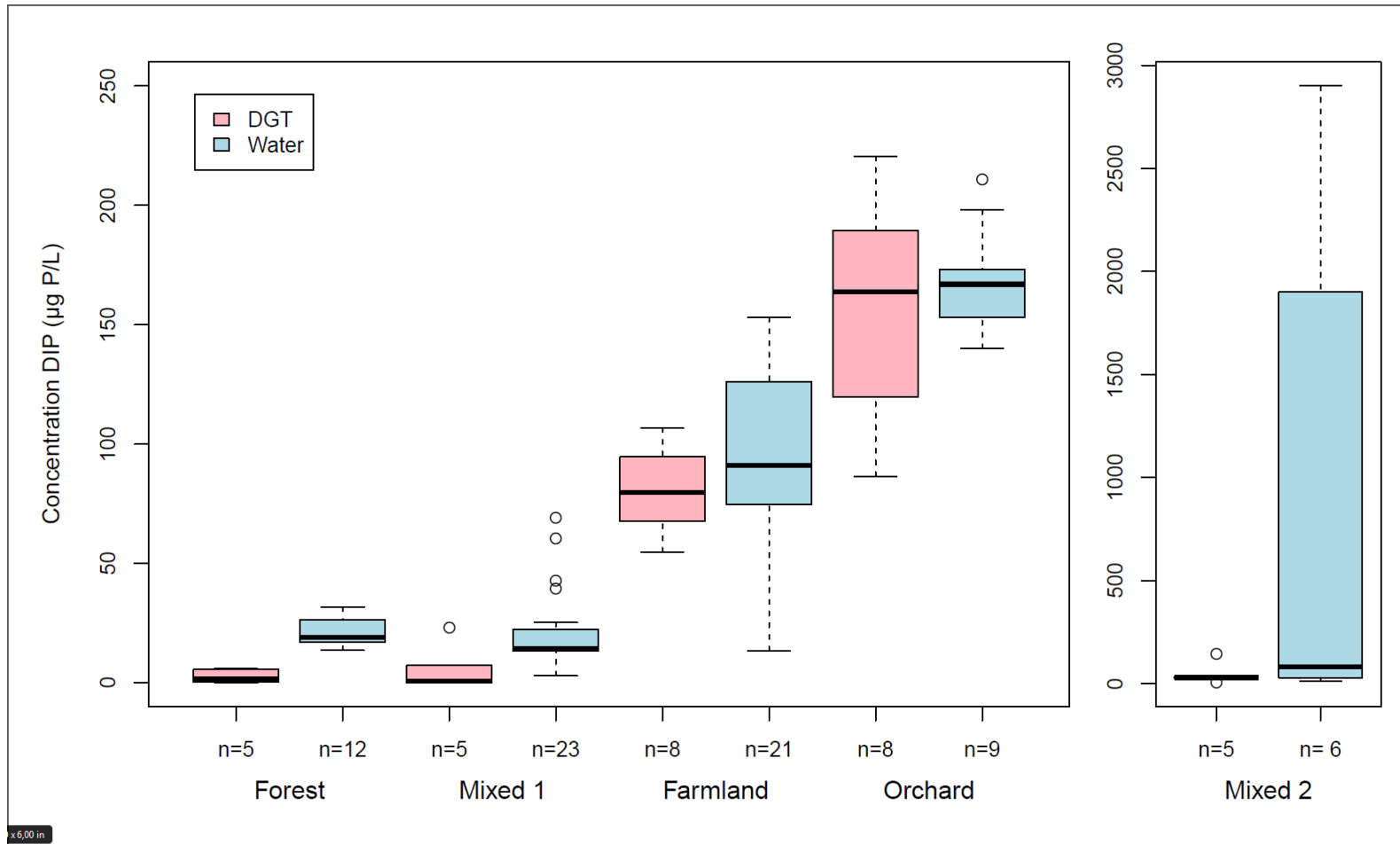
- **TP:** - Forest < Farmland < Mixed 2 < Orchard < Mixed 1
 - Same sequence in **soil**
 - Pettersen (2014) and
 - Joshi (2014)
- **PP:** - Related to loading of suspended solids
- **DOP:** - Relatively high in forest
 - DOM
- **DIP:** - Relatively high in orchard
 - Agricultural practises

DGT P fractions



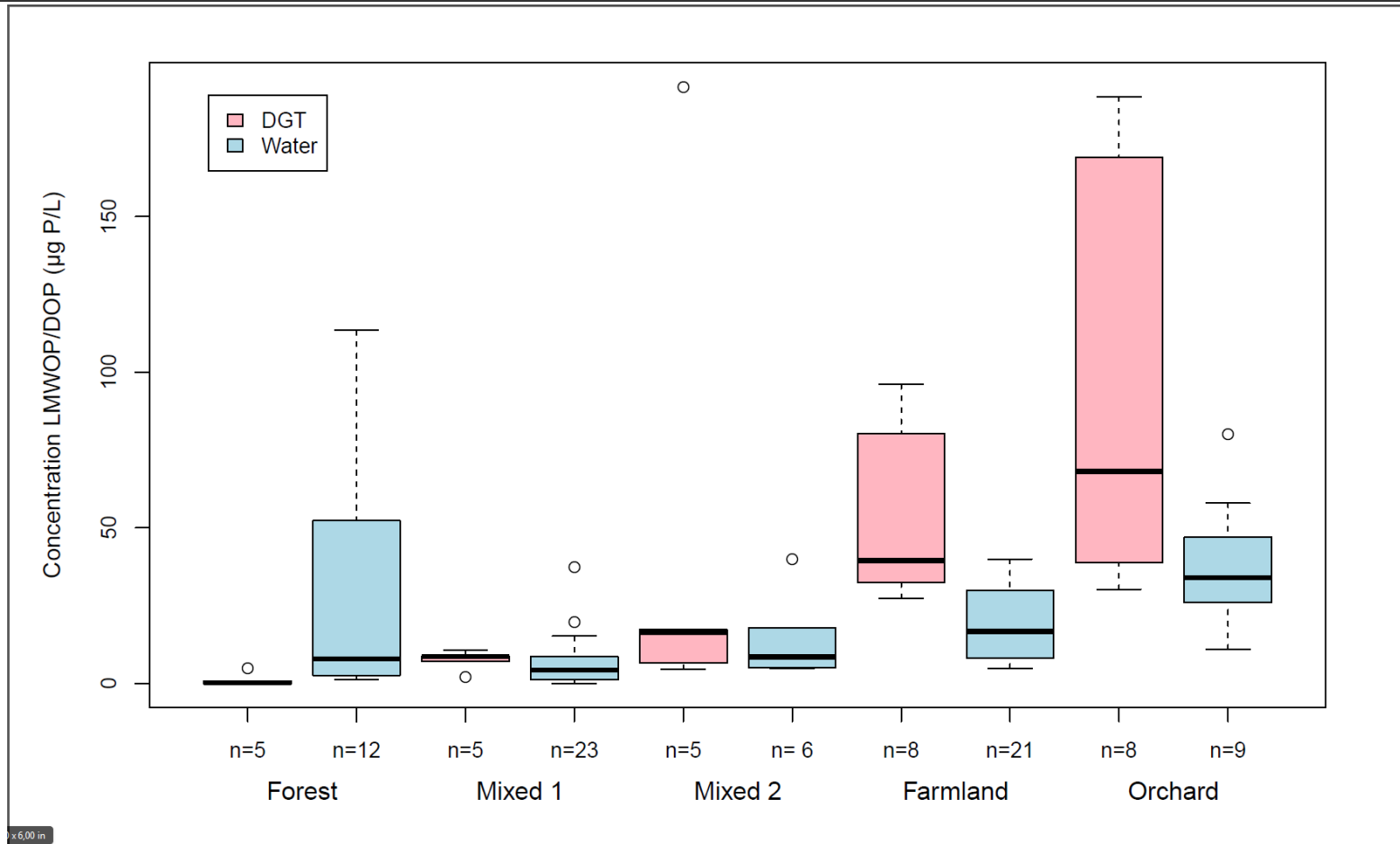
- **Rivers:** Similar P trend
 - Same as in water and soil
- **Fish ponds:** Difference in amount of TP
- **DIP:** Constitutes more than 50% of TDP in the rivers and fish ponds
- **Reservoir:** Surprise high concentration in middle depth

Water Vs DGT fraction – DIP fraction



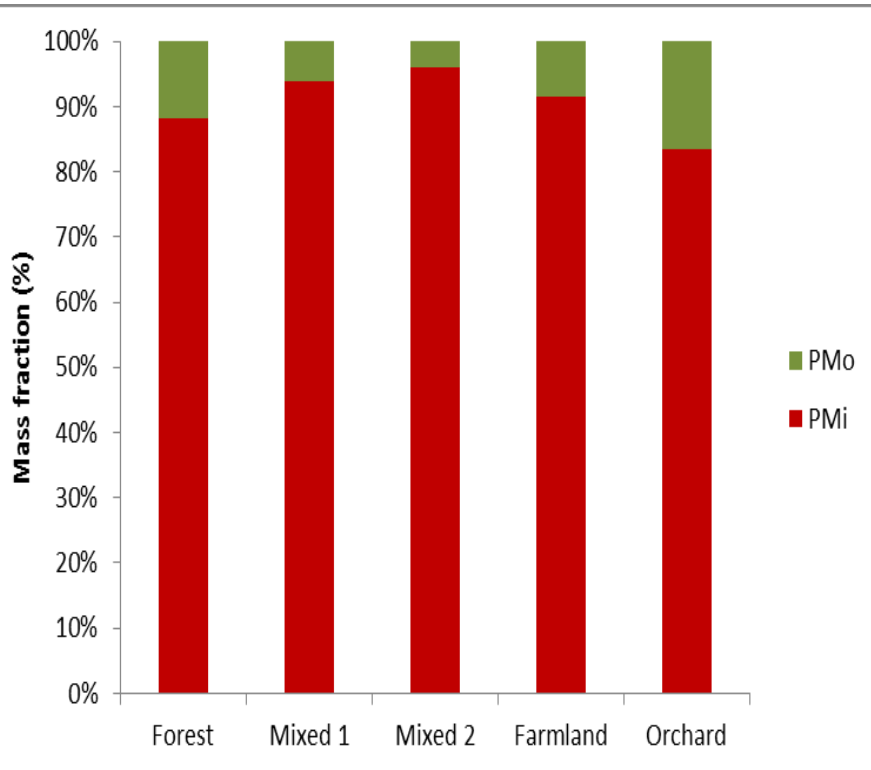
- **Results:** The two methods produce comparable results
- **Difference:** Methods (grab Vs Average) and hydrological fluctuations

Water Vs DGT fraction – DOP fraction



- **Results:** DGT-DOP higher than water-DOP (except in forest)
- **Discrepancy:** Errors due to value of diffusion coefficient used and LOD
- **Farmland/Orchard:** Difference due to large temporal variations (episodes)

Particulate characterization

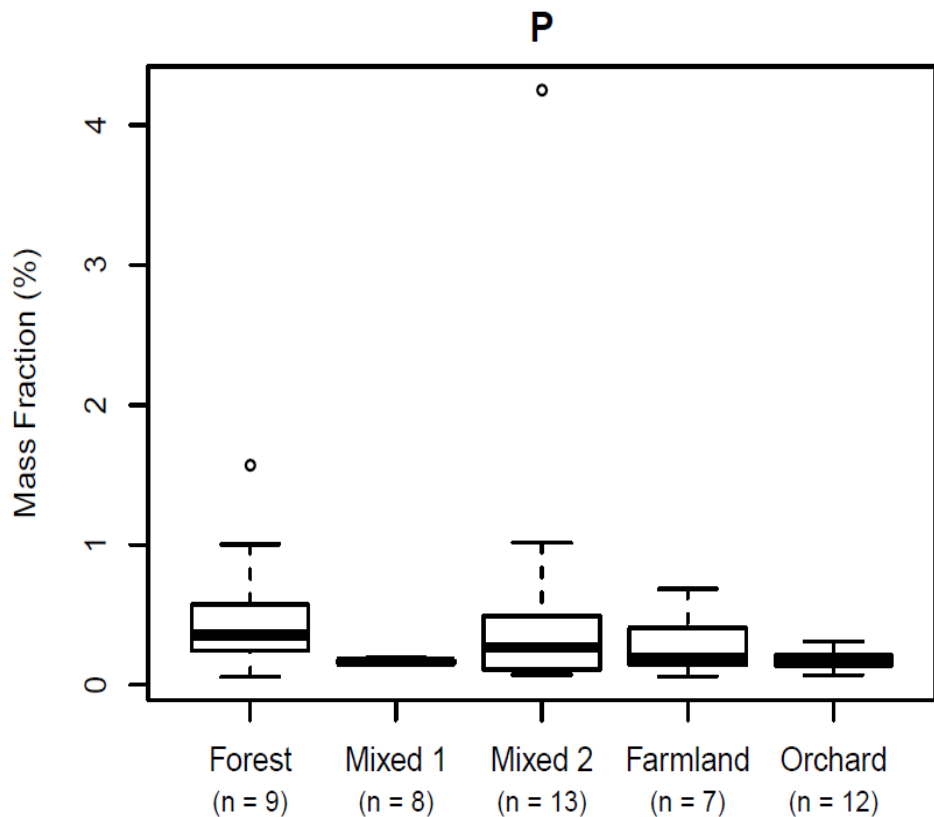


- Role of particles
 - Absorb or desorb P
- Different flow regimes
 - High flow
 - Low flow
 - Episodes

(July - September 2014)

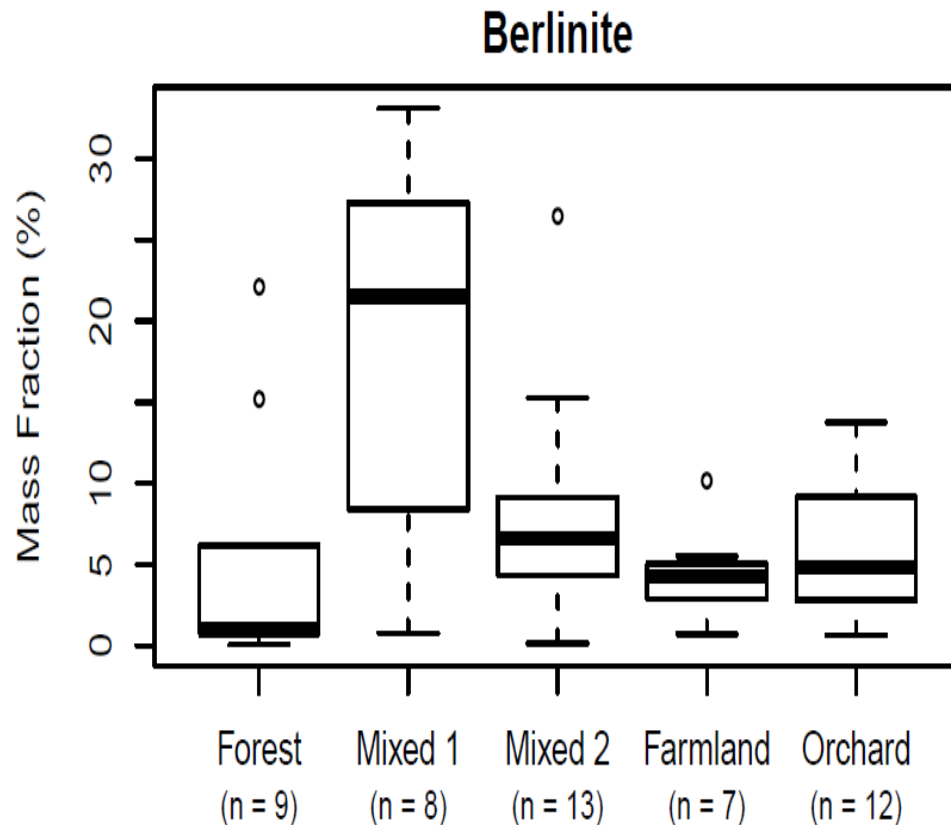
- Over **85%** content is inorganic

Particles – Elemental composition



- Main cations: - **Al** and **Ca**
- **Ca**: Lowest in the forest and orchards and highest in mixed water sheds
 - They have lower base saturation than farming land (Joshi, 2014)
- **P**: Different from river samples
 - Probably due to method used (MBM Vs ICP-OES)
 - Highest in forest and low in orchard – Difference in sorption index (PSI) which is highest in forest (Joshi, 2014)

Particles - Mineral composition



- No apatite and Vivianite
 - P likely from **anthropogenic** sources
- No clear mineral variation with land use and flow regimes
- Main mineral: **-1:1 clay**, in soil its quartz (Pettersen, 2014)
 - Easy of erosion
 - Clay – P mobility and transport
- Berlinite (AlPO_4)
 - Crystalline form
 - Industrial source

Conclusion

- Water chemistry is governed by Ca^{2+} , Mg^{2+} and HCO_3^-
- P precipitation is governed by Ca^{2+}
- Dominant P fraction in agricultural land is DIP
- DGT and grab water sampling are comparable (DIP)
- Al and Ca are the main elements in the eroded particles
- Eroded particles content is mainly clay(1:1) mineral
- Presence of Berlinite should be investigated further



Thank You!

